

A SYNTHESIS OF PROSTAGLANDIN  $F_{2\alpha}$  ( $PGF_{2\alpha}$ )

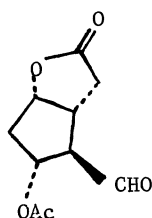
Tadao TANOUCHI, Seiji KORI, and Masaki HAYASHI\*

Research Institute, Ono Pharmaceutical Co., Ltd.,

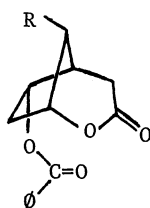
3-1-1, Sakurai, Shimamoto-cho, Mishima-gun, Osaka 618

$PGF_{2\alpha}$  15 was synthesized via the  $\delta$ -lactone aldehyde 2, which is much more stable than the Corey's aldehyde 1.<sup>1)</sup>

Although the  $\gamma$ -lactone aldehyde 1 has now been demonstrated to be one of the most versatile key intermediate for prostaglandin synthesis,<sup>1)</sup> the extreme thermolabilities for 1 made this route very difficult for the actual execution. We wish to report herein a new route to prostaglandins through the  $\delta$ -lactone aldehyde 2, which is much more stable and easy to handle.<sup>2)</sup>



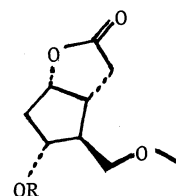
(1)



(2) R= CHO

(10) R=  $CH_2OCH_3$

(11) R=  $CH_2OH$



(3) R=  $COCH_3$

(4) R= H

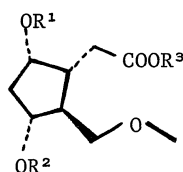
(5) R= THP

(THP= 2-tetrahydropyranyl)

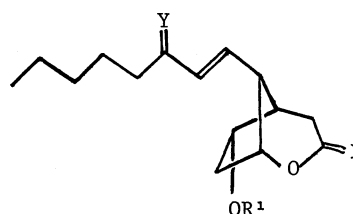
Successive treatment of the acetate 3 with (1) potassium carbonate in methanol, (2) dihydropyran and p-toluenesulfonic acid in methylene chloride, (3) 4N-sodium hydroxide in methanol and careful neutralization, (4) diazomethane in ethyl acetate ( $-78^\circ C$ ), (5) benzoyl chloride and pyridine in methylene chloride, afforded the acyl ester 7<sup>3)</sup> in 86% over-all yield.

Saponification of 7 with 1N-sodium hydroxide in methanol at room temperature gave the acid 8<sup>3)</sup> and the recovered acid 6<sup>3)</sup> in 44% and 36% yield, respectively.

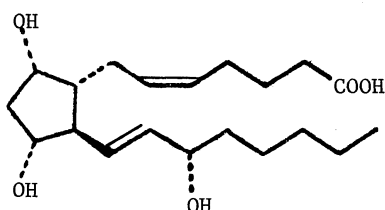
Hydrolysis of 8 with 1N-hydrochloric acid in tetrahydrofuran followed by lactonization of 9 with p-toluenesulfonic acid in benzene gave the  $\delta$ -lactone 10<sup>3),4)</sup> in 83% yield. The hydroxy compound 13 was obtained by successive treatment of 10 with (1) boron tribromide in methylene



- (6)  $R^1 = H$ ,  $R^2 = THP$ ,  $R^3 = H$   
 (7)  $R^1 = CO\emptyset$ ,  $R^2 = THP$ ,  $R^3 = CH_3$   
 (8)  $R^1 = CO\emptyset$ ,  $R^2 = THP$ ,  $R^3 = H$   
 (9)  $R^1 = CO\emptyset$ ,  $R^2 = H$ ,  $R^3 = H$



- (12)  $X = O$ ,  $Y = O$ ,  $R^1 = CO\emptyset$   
 (13)  $X = O$ ,  $Y = \begin{smallmatrix} H \\ \diagup \\ OH \end{smallmatrix}$ ,  $R^1 = CO\emptyset$   
 (14)  $X = \begin{smallmatrix} H \\ \diagup \\ OH \end{smallmatrix}$ ,  $Y = \begin{smallmatrix} H \\ \diagup \\ OH \end{smallmatrix}$ ,  $R^1 = H$



(15)

chloride, (2) Collins reagent in methylene chloride, (3) sodium hydride and dimethyl 2-oxoheptylphosphonate in tetrahydrofuran,<sup>5)</sup> (4) sodium borohydride in methanol, in 61% over-all yield from 10.

Diisobutylaluminum hydride reduction of 13 in toluene gave the hemiacetal 14<sup>3)</sup> (86% yield after column chromatography on silica gel), which was converted into  $PGF_2\alpha$  15<sup>6), 7)</sup> by Wittig reaction with 4-carboxy-n-butyldiene triphenylphosphorane (5 equiv.) in dimethyl sulfoxide in 26% yield after column chromatography on silica gel. 15-Epi- $PGF_2\alpha$  was obtained in 24% yield together with  $PGF_2\alpha$ .

#### Reference and Footnotes

- 1) E. J. Corey, T. K. Schaaf, W. Huber, U. Kolliker, and N. M. Weinshenker, J. Amer. Chem. Soc., 92 397 (1970).
- 2) The aldehyde 2 was stable after concentration by rotary evaporator (bath temp. 30°C) and standing overnight at +20°C. This stability will make industrial production of PGs very easy.
- 3) Ir and nmr spectra were in agreement with the assigned structure.
- 4) Ir absorption of  $1740\text{ cm}^{-1}$  shows  $\delta$ -lactone.
- 5) Only the trans isomer was obtained.
- 6) The 5,6-cis isomer was formed stereoselectively.
- 7) Ir and nmr spectra,  $R_f$  values on tlc and the biological activity were completely identical with those of the authentic sample.

(Received April 19, 1976)